

Physics 2305
Quiz 14—Form A

17 April, 2000

1. Copper has a specific heat of $386 \text{ J kg}^{-1} \text{ K}^{-1}$, and it melts at 1356 K . How much heat must a 1.0 kg sample initially at 315 K absorb to reach the melting point?

- A) 0.4 kJ C) 402 kJ
B) 122 kJ D) 523 kJ

2. A red giant star has a typical temperature of 2800 K and a radius of roughly 100 million km . How much power does it radiate? (Assume that it's a spherical blackbody.)

- A) $3.5 \times 10^6 \text{ W}$ C) $1.1 \times 10^{29} \text{ W}$
B) $3.5 \times 10^{12} \text{ W}$ D) $4.4 \times 10^{29} \text{ W}$

Equations and constants:

$$\sigma = 5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$$

$$\Delta L = L \alpha \Delta T \quad \Delta V = V \beta \Delta T \quad \beta = 3\alpha$$

$$Q = c m \Delta T \quad Q = L m$$

$$P_r = A \varepsilon \sigma T^4 \quad H = A \Delta T / \Sigma R \quad R = L / k$$

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Quiz 14—Form B

17 April, 2000

A sample of ice is cooled from 0.0°C to -40.0°C . Its initial mass and volume are 7.34 g and 8.00 cm^3 .

1. Ice has an expansion coefficient of $5.1 \times 10^{-5}\text{ K}^{-1}$. What is its final volume?

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|-----------------------|-----------------------|
| A) 0.02 cm^3 | C) 7.95 cm^3 |
| B) 0.05 cm^3 | D) 7.98 cm^3 |

2. How much heat was removed from the ice? The specific heat of ice is $2220\text{ J kg}^{-1}\text{ K}^{-1}$.

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|-------------------|--------------------|
| A) 16 J | C) 16 kJ |
| B) 650 J | D) 650 kJ |

Equations and constants:

$$\sigma = 5.67 \times 10^{-8}\text{ W m}^{-2}\text{ K}^{-4}$$

$$\Delta L = L \alpha \Delta T \quad \Delta V = V \beta \Delta T \quad \beta = 3\alpha$$

$$Q = c m \Delta T \quad Q = L m$$

$$P_r = A \varepsilon \sigma T^4 \quad H = A \Delta T / \Sigma R \quad R = L / k$$

Physics 2305
Quiz 14—Form C

17 April, 2000

For both problems, assume we are dealing with a penny made of 2.5 g of pure copper.

$$L_F = 207 \text{ kJ/kg} \quad \alpha = 1.7 \times 10^{-5} \text{ K}^{-1}$$
$$c = 0.386 \text{ kJ kg}^{-1} \text{ K}^{-1}$$

1. If a penny is heated to the melting point (1356 K), how much *additional* heat is needed to melt it?

- A) 83 J C) 1200 J
B) 520 J D) 1300 J

2. A penny at room temperature (290 K) has a diameter of 19.05 mm. How hot would it have to be to expand by 1%?

- A) 490 K C) 880 K
B) 590 K D) $T_f > 1356 \text{ K}$

Equations and constants:

$$\sigma = 5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$$

$$\Delta L = L \alpha \Delta T \quad \Delta V = V \beta \Delta T \quad \beta = 3\alpha$$

$$Q = c m \Delta T \quad Q = L m$$

$$P_r = A \varepsilon \sigma T^4 \quad H = A \Delta T / \Sigma R \quad R = L / k$$