

The *JWST* Proto-PAH project. Computational modeling of the emission carriers

A. RICCA,^{1,2} G. C. SLOAN,^{3,4} E. PEETERS,^{5,6} J. CAMI,^{5,6} N. CLARK,^{5,6} M. MATSUURA,⁷ R. SAHAI,⁸
J. BERNARD-SALAS,^{9,10} D. A. GARCÍA-HERNÁNDEZ,^{11,12} J. LI,^{11,12} AND C. BHATT^{5,6}

¹*NASA Ames Research Center, MS 245-6, Moffett Field, CA 94035-1000, USA*

²*Carl Sagan Center, SETI Institute, 339 Bernardo Avenue, Suite 200, Mountain View, CA 94043, USA*

³*Space Telescope Science Institute, 3700 San Martin Drive, Baltimore, MD 21218, USA*

⁴*Department of Physics and Astronomy, University of North Carolina, Chapel Hill, NC 27599-3255, USA*

⁵*Department of Physics and Astronomy, The University of Western Ontario, London, ON N6A 3K7, Canada*

⁶*The Institute for Earth and Space Exploration, The University of Western Ontario, London, ON N6A 3K7, Canada*

⁷*Cardiff Hub for Astrophysics Research and Technology (CHART), PACES, Cardiff University, The Parade, Cardiff CF24 3AA, UK*

⁸*Jet Propulsion Laboratory, California Institute of Technology, 4800 Oak Grove Dr., Pasadena, CA 91109, USA*

⁹*ACRI-ST, Centre d'Etudes et de Recherche de Grasse (CERGA), 10 Av. Nicolas Copernic, 06130 Grasse, France*

¹⁰*INCLASS Common Laboratory, 10 Av. Nicolas Copernic, 06130 Grasse, France*

¹¹*Instituto de Astrofísica de Canarias, C/ Via Láctea s/n, E-38205 La Laguna, Spain*

¹²*Departamento de Astrofísica, Universidad de La Laguna (ULL), E-38206 La Laguna, Spain*

(Received XXX, 2024; Revised XXX, 2024)

Submitted to ApJ

ABSTRACT

The infrared spectra of many carbon-rich post-asymptotic giant branch stars are dominated by emission features from aromatic and aliphatic hydrocarbons, but the chemical structure of the carriers remains unidentified. Class D sources show unusual emission profiles with strong aliphatic emission, providing a stringent test of carrier candidates. Here we investigate whether hydrogenated carbonaceous molecular particles can be the carriers of these features. Rather than assuming candidate geometries we generate the initial structures using ab initio molecular dynamics simulations and optimize them using density functional theory. We then compare the computed emission spectra to *JWST* NIRSpec and MIRI/MRS observations of Class D sources. The best-fitting structures contain sizeable domains with defect-bearing aromatics and hydrogenated-aromatics connected by aliphatic bridges, in contrast to the “aromatic” cluster model of isolated two- to three-ring aromatics connected by aliphatic and olefinic bridges. As a representative example, we discuss in detail the molecular particle C₁₃₀H₉₆ (radius of 5.5 Å) and compare its computed spectrum to the *JWST* spectra of IRAS 05110-6616, a Class D2 source with apparent shifts of the aromatic C—H out-of-plane bands. The calculated spectra are in qualitative agreement with the observed 3 μm profile, the aliphatic C—H bending bands, the broad 8 μm feature, the 11–14 μm C—H out-of-plane region, and the bands at 16 and 21 μm. We hypothesize that UV photo-driven fragmentation of such molecular particles releases aromatics linked by aliphatics, followed by the formation of polycyclic aromatic hydrocarbons and fullerenes as these objects evolve into planetary nebulae.

Keywords: Infrared astronomy — Molecular spectroscopy — Theoretical techniques

1. INTRODUCTION

Carbon-rich stars are important sources of carbonaceous materials in the ISM. After ejection, the material is further processed by UV radiation, energetic particles and strong shocks and ends with its incorporation into newly formed stars and planetary systems. Under-

standing the currently unknown steps along this path and elucidating the chemical structures of the emission carriers is of key importance. It is generally thought that photo-processing can change the ratio of aliphatic carbon (present in chains or rings, e.g., sp^3 hybridized C) versus aromatic carbon (present in aromatic rings, e.g., sp^2 hybridized C).

In some benign environments associated with post-asymptotic giant branch (post-AGB) stars and proto-planetary nebulae (PPNe), a relatively high aliphatic fraction has been observed in infrared (IR) emission spectra. Aliphatic bands peak at $3.4\text{--}3.5\text{ }\mu\text{m}$ (Geballe & van der Veen 1990; Geballe et al. 1992, 1994) and at 6.9 and $7.25\text{ }\mu\text{m}$ in addition to broad features near $8\text{ }\mu\text{m}$ and in the $11\text{--}15\text{ }\mu\text{m}$ region (Sloan et al. 2007, 2014; Sloan 2017; Matsuura et al. 2014; Jensen et al. 2022). In harsh environments associated with HII regions, Herbig AeBe stars, reflection nebulae (RNe), diffuse ISM, and planetary nebulae (PNe), the fraction of C atoms in aliphatic form observed in the IR emission spectra is much lower (10%; e.g., Li & Draine 2012; Yang et al. 2016). These spectra are classified as Class A and B (Peeters et al. 2002). In these objects, aromatic bands at 6.2 , 7.7 , 8.6 , 11.2 and $12.7\text{ }\mu\text{m}$, generally attributed to polycyclic aromatic hydrocarbons (PAHs), dominate the spectra, while aliphatic bands at 6.9 and $7.25\text{ }\mu\text{m}$ are negligible. Class C sources include Young Stellar Objects (YSOs; Acke et al. 2010) and post-AGB stars (Peeters et al. 2002; Sloan et al. 2007). Their emission spectra contain broader features than Class A and B spectra and the aliphatic/aromatic content is in-between that of Classes A/B and D sources (Sloan et al. 2007; Pino et al. 2008). Class D sources have been further divided as Class D1 and D2 (Matsuura et al. 2014; Sloan et al. 2014). Their emission carriers, which we will refer here as ProtoPAHs, have spectra that are inconsistent with PAH spectra (Clark et al. 2026, submitted). The chemical structure of the carriers remains elusive.

JWST program ID 4678 (PI: G. C. Sloan) observed a sample of seven targets, all of them carbon-rich post-AGB objects in the Large Magellanic Cloud (LMC) and the Small Magellanic Cloud (SMC). The sample includes two sources with Class D1 PAH spectra, two D2, two sources known or suspected to have strong absorption spectra, and one Class B source complemented with SMP LMC 058, a PN with Class B PAH emission that was observed as a calibration target by *JWST*. Class D observed spectra provide the observational constraints needed to characterize ProtoPAHs (Sloan et al. 2026, submitted).

The infrared emission of mixed aromatic/aliphatic carriers has previously been modeled with density func-

tional theory, using hand-constructed structures ranging from PAHs with aliphatic side groups (Sadjadi et al. 2015, 2017) to mixed aromatic/aliphatic organic nanoparticles (MAONs; Kwok & Zhang 2011; Kwok & Zhang 2013; Sadjadi et al. 2016).

Jones, A. P. & Ysard, N. (2025) used THEMIS (The Heterogeneous dust Evolution Model for Interstellar Solid; Jones 2013) to generate a three-dimensional “aromatic” cluster model, consisting of isolated clusters of two to three six-membered ring aromatics connected by aliphatic and olefinic bridges (Jones 2012; Micelotta et al. 2012). They further assessed the photon-driven fragmentation and Coulomb fragmentation of the “aromatic” cluster model. Alternatively, the molecular particles can be computed by starting from its precursor material, namely amorphous carbon (a-C). Computational ab initio molecular dynamics (AIMD) methods provide realistic models of a-C for densities spanning 0.95 to 3.5 g cm^{-3} (Bhattarai et al. 2018; Bauschlicher & Lawson 2010). The structures vary from interconnected wrapped and defective sp^2 sheets at 0.95 g cm^{-3} to almost tetrahedral (diamond-like) structures at 3.5 g cm^{-3} (Bilek et al. 2000).

In this work we seek to characterize the ProtoPAH emitters without prior assumptions about their structure by using a-C as a starting point and the constraints provided by *JWST* observations. We should note that these structures are not being proposed as carriers of the general AIB spectrum, but only as carriers of the specific sources that show a spectrum that is markedly different from the regular AIBs, such as the proto-PAH targets.

This paper is organized as follows. Section 2 describes the theoretical methods employed in this paper. Section 3 provides an analysis of the infrared (IR) spectral characteristics and structural properties of the molecular particle. Further astrophysical implications for these findings are discussed in Section 4, and Section 5 summarizes these results.

2. METHODS

The steps involved in generating the hydrogenated carbonaceous molecular particle structures, with a net charge of zero, are shown in Figure 1. Starting from a cubic box containing carbon atoms placed randomly we obtained bulk a-C. We generated a sphere out of the amorphous carbon cube and then hydrogenated it. Linear chains containing carbon atoms doubly coordinated (C_{sp}) were removed and long aliphatic chains were folded to allow ring formation. The geometry of the sphere was subsequently optimized.

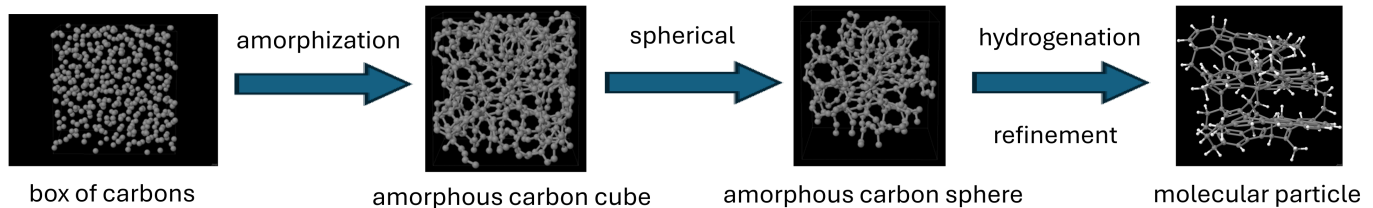


Figure 1. Steps involved in generating the hydrogenated carbonaceous molecular particle structures.

Bulk a-C was computed using ab initio molecular dynamics (AIMD) simulations and employing the Vienna ab initio simulation package (VASP), version 5.4.4.18Apr17 (Kresse & Hafner 1993, 1994; Kresse & Furthmüller 1996; Kresse & Furthmüller 1996). The Perdew-Wang 91 (PW91) exchange-correlation functional (Perdew & Wang 1992) and a plane wave basis set, in conjunction with the projector augmented wave method (PAW; Kresse & Joubert 1999), were employed. An energy cutoff of 400 eV for the plane-wave basis set and a $1 \times 1 \times 1$ grid of k points were used in all simulations.

We generated a cubic cell containing 128 carbon atoms with a density of 2.0 g cm^{-3} . The atoms were distributed randomly in the box with the constraint that their positions be at least 0.5 Å away from the edges and that the distance between the carbon atoms be greater than 1.3 Å . The cubic cell was then replicated along the x,y,z directions to produce a bulk structure. The atomic positions in the cell were relaxed at 0 K while keeping the dimensions of the cell fixed.

The amorphization step, shown in Figure 1, was performed following the procedure described by Bauschlicher & Lawson (2010) using three steps: (i) melting consisting in heating and equilibrating at 5000 K over 0.6 ps; (ii) quenching from 5000 K to 1 K over 4 ps and (iii) energy minimization at 0 K.

From the resulting a-C structure we extracted a sphere with a radius of 7 Å that captured enough aliphatic, aromatic, and olefinic (non-aromatic C=C double bonds) components. The bond connectivity was assessed and defects (carbon atoms with less than four bonds) were hydrogenated.

The geometries of the neutral molecular particles obtained from the MD simulations were reoptimized using the density functional theory (DFT) method, the 6-31G* basis set (Frisch et al. 1984) and their harmonic frequencies computed using the Gaussian 16 suite of programs (Frisch et al. 2016). We ensured that the structures had a minimum on the potential energy surface, i.e. they didn't produce imaginary frequencies.

The harmonic frequencies were scaled to lower values using three scaling factors, namely 0.959 for C-H stretches in the 3 μm region, 0.9715 for the 5-

10 μm region, and 0.9821 for everything above 10 μm (Bauschlicher et al. 2018). The frequencies (in cm^{-1}) and band intensities (in km mol^{-1}) were synthesized into emission spectra using an emission cascade model. Emission spectra were computed for the absorption of a single photon using excitation in the solar radiation field characterized by a 5770 K blackbody ($\approx 3.5 \text{ eV}$), and the entire cooling cascade was taken into account (see e.g., Boersma et al. 2010; Boersma et al. 2011; Boersma et al. 2014; Bauschlicher & Ricca 2010). The integration was done over wavenumbers (units of erg) and convolution with a line profile gave units of radiant energy in erg/cm^{-1} . Gaussian line profiles were used with full width at half maximum (FWHM) values of 30 cm^{-1} for bands shortward of 9 μm and 25 cm^{-1} for bands between $9\text{--}20 \text{ μm}$, based on comparisons with observations.

The geometries and vibrational normal modes were visualized using the software Jmol¹. All the structural and spectroscopical data will be made publicly available in the NGdb – Theoretical database².

3. RESULTS

3.1. Structural properties

We obtained seventy structures and assessed their relevance based on comparisons of their modeled emission spectra with *JWST* observations. Poor fits were obtained when (i) olefinic double bond chains linked aromatics and hydrogenated aromatics (i.e. aromatics with additional hydrogen atoms) and (ii) when only a very small number of aromatic rings were present in the domains. The best fits were obtained for molecular particles containing sizeable aromatic/hydrogenated aromatic domains (bright colors) connected by aliphatic links (grey color), as shown in Figure 2. A molecular particle representative of this class is $\text{C}_{130}\text{H}_{96}$ with a radius of 5.5 Å . $\text{C}_{130}\text{H}_{96}$ itself is not a unique astronomical carrier. Rather, it serves as a representative example demonstrating that a single molecular particle containing aromatics and aliphatics can account for several features that have previously been difficult to explain

¹ <https://jmol.sourceforge.net/>

² <https://nanograin.odr.io/>

simultaneously, and that such a structure can indeed produce many of the observed characteristics. This particular structure with its curved and defective ring systems, hydrogenated aromatic regions, and aliphatic connections may provide clues to how such particles could later fragment or rearrange into PAHs and fullerenes. In a subsequent paper we plan to extend this study to an ensemble of structures and include distributions in size, H/C, and C_{sp^2}/C_{sp^3} ratios.

For each domain, the breakdown of aromatic and hydrogenated aromatic rings, the total number of connections to the rings and the number and type of aliphatic links are given in Table 1. Most of the fully aromatic rings are six-membered rings whereas the hydrogenated aromatic rings contain five-, six- and seven-membered rings. Most of the aliphatic C–H links contain up to two carbons, with a few links containing three and four carbons. The number of aromatic carbons is 67, olefinic carbons is 16 and aliphatic carbons is 47. Twelve

aliphatic hydrogens are in 4 CH₃ groups, 46 aliphatic hydrogens are in 23 CH₂ groups, and 16 aliphatic hydrogens are in 16 CH groups. Together with the 15 aromatic and 7 olefinic H atoms, this gives a total of 96 H atoms. This mixed character is due to the presence of defects, such as pairs of five/seven-membered rings, which reduce the number of aromatic bonds and create carbon radicals (with less than four bonds) which can react with hydrogen atoms to form hydrogenated aromatic rings and/or with other carbon radicals to form C–C bonds. The presence of five-membered rings induces curvatures in the structure that depend on the number of five-membered rings and their relative positions. The rings are linked by aliphatics attached at their edges. The structure is more strained and disordered than PAHs linked by aliphatic chains. It can be viewed as an intermediate component between a-C and PAHs.

Table 1. Breakdown of the number of aromatic and hydrogenated aromatic rings, along with the total number of carbon links attached to each domain (total C-links), and the type and number of aliphatic links (CH₃, CH₂, 2(CH₂) (2C), CHCHCH₃ (3C-1), CHCH₂CH (3C-2), CHCH₂CH₂ (3C-3), CH₂CHCH₂CH₂ (4C)).

Domain	Arom. rings			Hydr. arom. rings			Total C-links	Aliphatic links						
	Five	Six	Seven	Five	Six	Seven		CH ₃	CH ₂	2C	3C-1	3C-2	3C-3	4C
Red	2	1	1	2	1	1	5		2	1	1			
Green		5		1		1	8		2	4	1			
Magenta		1		1	1		6	1		3		1		1
Cyan		6		1			6	1		1		1	1	1
Blue		1					4	1	1	1				
Yellow				1	1		6							

3.2. Emission spectra

NIRSpec and MIRI/MRS spectra of the Class D2 IRAS 05110-6616 (IRAS 05110 hereafter) source (Sloan et al. 2026, submitted) are shown in Figure 3 along with the computed emission spectra of the neutral molecular particle C₁₃₀H₉₆, for the absorption of a 3.5 eV photon. The computed band positions, intensities, and assignments are given in Table 2.

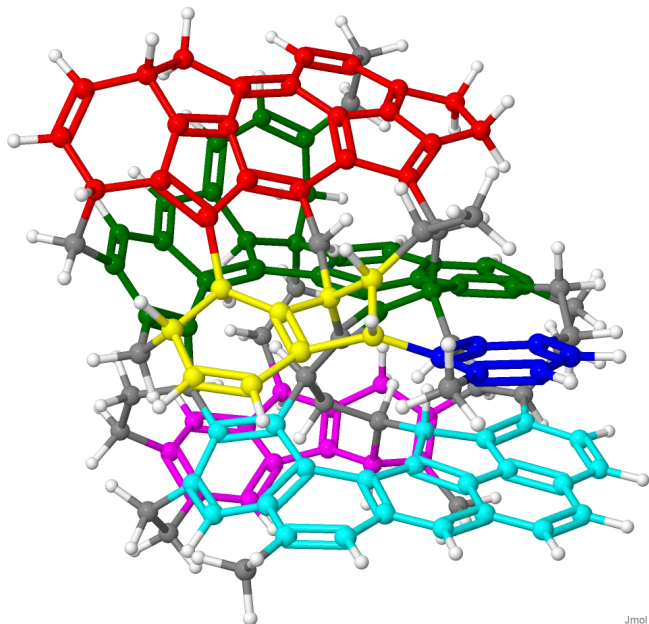
A detailed analysis of the NIRSpec spectrum of IRAS 05110 was performed using gaussian fits of the normalized, continuum-subtracted emission bands (Clark et al. 2026, submitted). It showed that the main emission bands in the NIRSpec range can be decomposed into

three features: *i*) a wide aromatic C–H stretching band at 3.284 μ m with some olefinic C–H stretching contributions at 3.24 and 3.35 μ m; *ii*) a CH₂ asymmetric stretching band at 3.425 μ m with a shoulder at 3.38 μ m due to a CH₃ asymmetric stretching band; *iii*) an aliphatic band at 3.489 μ m with contributions from a CH₃ symmetric stretching band (3.4838 μ m) and a CH₂ symmetric stretching band (3.5075 μ m). No strong band was observed at 3.40 μ m which indicates that superhydrogenated aromatics, such as 1,2,3,6,7,8-hexahdropyrene (Demyk et al. 2026), are not present in substantial amounts.

The computed 3 μ m emission (Figure 3) contains three bands that fall at wavelengths comparable to observations: *i*) an aromatic C–H stretching band at 3.278 μ m with a small olefinic C–H stretching band contribution

Table 2. Summary of computed band positions (μm), intensities (kcal mol^{-1}), and assignments for $\text{C}_{130}\text{H}_{96}$.

Band position	Intensity	Assignment
3.278	0.023737	aromatic C–H stretching
3.424	0.087557	CH_2 asymmetric stretching in aliphatic links and CH_3 asymmetric stretching
3.50	0.02158	CH_2 symmetric stretching in hydrogenated aromatics and CH_3 symmetric stretching
6.261	0.797	C–C aromatic stretching
6.623	1.10	C–H in-plane bending and CH_2 scissoring in long aliphatic links
6.813	1.72	CH_2 in-plane scissoring in short aliphatic links
7.196 (blended)	–	CH_3 symmetric bending (umbrella mode)
7.372 – 9.25	9.78	aliph., arom., olef. in-plane C–H bending, skeletal C–C distortions
10.961	3.99	aromatic CH_{oop} solo and CH_2 in-plane rocking
11.242	5.34	aromatic CH_{oop} solo
11.783	2.03	aromatic CH_{oop} solo and CH_2 in-plane rocking
12.444	3.82	aromatic CH_{oop} duo
13.371	2.98	CH_2 in-plane rocking and CH_{oop} solo
16.16	3.70	skeletal bending
18.63	2.79	skeletal bending
20.70	2.22	skeletal bending

**Figure 2.** Structure of the hydrogenated carbonaceous molecular particle $\text{C}_{130}\text{H}_{96}$ with the various aromatic/hydrogenated aromatic domains, highlighted in different colors, connected by aliphatic links, shown in grey.

sistent with the absence of superhydrogenated aromatic domains. The ratio $3.4 \text{ aliphatic} / \sum(3.3, 3.4)$ is 0.82, which is slightly larger than the value of 0.7749 from IRAS 05110 (Clark et al. 2026, submitted). The bands at 3.278 and 3.50 μm have comparable intensities and are a factor of 3.9 smaller than the 3.424 μm band. The larger intensity of the 3.424 μm band is consistent with the structure containing more CH_2 groups in aliphatic links than in hydrogenated aromatics. Despite a larger CH_2/CH_3 ratio than observed for IRAS 05110 the overall computed 3 μm profile is in qualitative agreement with the *JWST* profile. We should note that large structures with 150 carbon atoms or more are expected to produce very weak 3 μm bands for the absorption of a 3.5 eV because the absorbed energy is distributed over more vibrational modes in a larger particle.

The MIRI/MRS spectrum contains several distinct features: *i*) an aromatic C–C stretching band at 6.25 μm ; *ii*) a CH_2 asymmetric bending band and a CH_3 asymmetric bending band at 6.90 μm ; *iii*) a CH_3 symmetric bending band at 7.26 μm ; *iv*) a broad feature extending from ≈ 7.4 to 9 μm of unknown origin; *v*) three C–H out-of-plane bending (OOP) bands at 11.39, 12.28 and 13.23 μm with decreasing intensities relative to each other (Clark et al. 2026, submitted). In addition, in the far-IR the MIRI/MRS spectrum contains bands at 15.88, 17.0 and the “21.0 μm ” feature (Sloan et al. 2026, submitted).

The computed 5.0–13.7 μm emission (Figure 3) shows that the intensities of the bands in the 6–10 μm are smaller than those of OOP bands (11–14 μm region) by approximately a factor of 3, consistent with neutral

at 3.267 μm ; *ii*) a CH_2 asymmetric stretching band in aliphatic links at 3.424 μm with an inflection point at 3.38 μm due to a CH_3 asymmetric stretching mode and a blue rise starting at 3.32 μm due to olefinic C–H and CH_2 stretching modes; *iii*) a band at 3.50 μm due to a CH_2 symmetric stretching mode in hydrogenated aromatics and a CH_3 symmetric stretching mode. Our spectra do not produce a strong peak at 3.40 μm con-

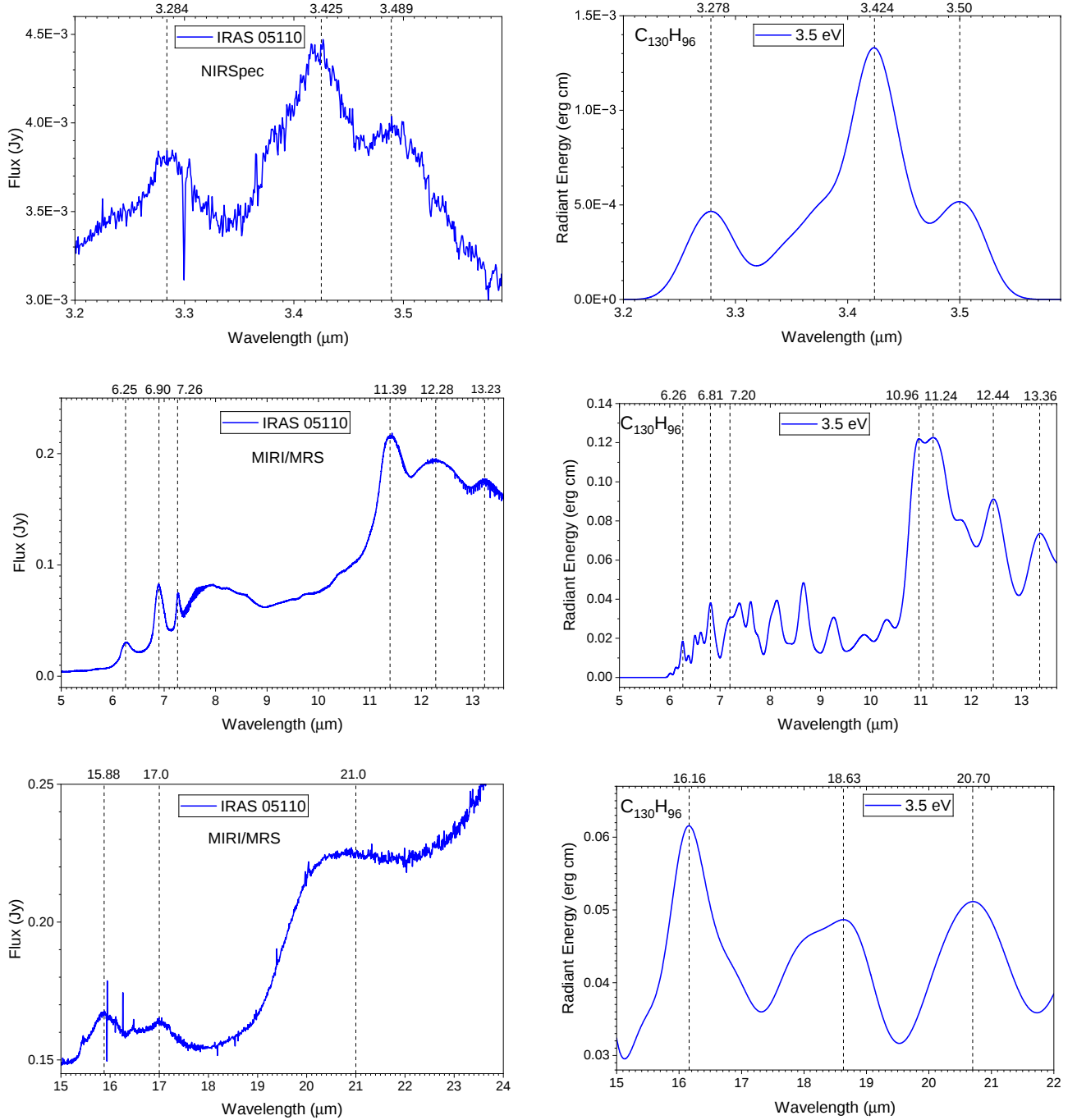


Figure 3. Comparison of *JWST* spectra of IRAS 05110 (left) with computed emission spectra for the absorption of a 3.5 eV photon (right). Dotted vertical lines indicate band positions.

335 molecules (Langhoff 1996). The aromatic C–C stretch-
 336 ing peaks at 6.26 μm in agreement with observations,
 337 and a weak olefinic C–C stretching band is visible at
 338 6.12 μm . The position of the CH_2 scissoring mode is
 339 dependent on the level of molecular strain. In short
 340 aliphatic links, it peaks at 6.811 μm and in hydrogenated
 341 aromatic rings it peaks around 6.9 μm (Sandford et al.
 342 2013; Materese et al. 2017). For long aliphatic links with

343 low strain the scissoring mode blueshifts to 6.67 μm . Ta-
 344 ble 1 shows that our structure contains more CH_2 groups
 345 in short aliphatic links than in hydrogenated aromatics,
 346 which is reflected in the CH_2 scissoring mode peak-
 347 ing at 6.81 μm . A few weaker CH_2 scissoring modes
 348 at 6.6 μm are due to aliphatic links containing three
 349 and four carbons. In observations, the majority of emis-
 350 sion is centered at 6.90 μm , with only a small contribu-

tion at $6.81\ \mu\text{m}$, and has been attributed to an aliphatic C–H scissoring mode adjacent to an aromatic ring (Clark et al. 2026, submitted). As the position of the $6.9\ \mu\text{m}$ band is sensitive to the specific chemical structure it is possible that the discrepancy between our calculations and observations is due to the specific structure of our molecular particle. The computed CH_3 symmetric bending mode (i.e. umbrella mode) peaks at $7.20\ \mu\text{m}$, at a slightly shorter wavelength than the observed band, and it is followed by a series of peaks attributed to combinations of aliphatic, aromatic, and olefinic C–H bending modes that are coupled with the collective skeletal C–C modes. Given the disordered and distorted nature of these molecular particles and the fact that multiple structures are likely to be present, it is expected that many bands from different species will overlap and form a broad feature in this range (Lazzarini et al. 2016) as seen in the IRAS 05110 MIRI/MRS spectrum.

Two weak computed olefinic OOP modes peak at 10.34 and $10.85\ \mu\text{m}$ and a weak CH_2 in-plane rocking band peaks at $9.87\ \mu\text{m}$. The “ $11.2\ \mu\text{m}$ ” feature is broad and consists of a band at $10.96\ \mu\text{m}$, attributed to a combination of an aromatic C–H solo OOP mode and an aliphatic CH_2 in-plane rocking mode (in aliphatic link and hydrogenated aromatic), and a band at $11.24\ \mu\text{m}$ assigned to an aromatic C–H solo OOP band. This broadening is consistent with the modification of OOP bands by the presence of aliphatics (Sandford et al. 2013; Dar-
tois et al. 2020). The band at $12.44\ \mu\text{m}$ is attributed to an aromatic C–H duo OOP mode, and the band at $13.36\ \mu\text{m}$ is due to a combination of an aliphatic CH_2 in-plane rocking and an aromatic C–H solo OOP mode. The interaction of aromatics and aliphatics explains the shifting of the bands observed in the Class D2 spectra.

In the far-IR we obtained three bands at 16.16 , 18.63 , and $20.70\ \mu\text{m}$ due to skeletal bending modes. The computed band positions at 16.16 and $20.70\ \mu\text{m}$ are in agreement with observations whereas the band at $18.63\ \mu\text{m}$ is redshifted. It is quite possible that similar species with slightly different structures can produce an entirely different skeletal mode instead of the one at $18.63\ \mu\text{m}$ as the skeletal modes are sensitive to the whole vibrational structure. We should stress that other carbonaceous particles are likely to contribute to the 21 and $30\ \mu\text{m}$ bands (see e.g., Papoular 2013; Volk et al. 2020) and our molecular particle is another example of a possible candidate.

Overall, our computed spectra show all the general features observed for IRAS 05110 and one can envision that an ensemble of hydrogenated carbonaceous molecular particles could provide an even better fit.

4. DISCUSSION

Our computed molecular particle is a representative example demonstrating that a single molecular particle containing aromatics and aliphatics can account for several features that have previously been difficult to explain simultaneously and reproduce many of the observed emission characteristics. It supports the hypothesis by Clark et al. (2026, submitted) that the emission, in a small subset of Class D carriers, is due to a single component instead of two coexisting components. They based their hypothesis on the position of the $6.9\ \mu\text{m}$ feature and the perturbation of the aromatic OOP modes which suggested that the aromatic and aliphatic groups are close to one another within the same carrier. The computed structure of our representative component contains aromatic/hydrogenated aromatic domains, with pairs of five-/seven-membered ring defects, connected by aliphatic links. It is different from the three-dimensional “aromatic” cluster model, which is less strained and contains only small defect-free aromatic domains without the presence of hydrogenated aromatics.

During the transition from the post-AGB phase to PNe, the molecular particles will experience harsher conditions and UV photon-driven fragmentation. Wang (2017) studied the thermal decomposition of n-hexene (a partially hydrogenated benzene) and n-hexane (a fully hydrogenated benzene). The resulting bond (marked by a λ) energies were $3.2\ \text{eV}$ for an allyl-adjacent group, $-\text{C}-\lambda-\text{C}=\text{C}-$, present in n-hexene, $3.8\ \text{eV}$ for an aliphatic $-\text{C}-\lambda-\text{C}-\lambda-\text{C}-\lambda-\text{C}-$ group present in n-hexane, $4.3\ \text{eV}$ for a C–H bond, and $4.4\ \text{eV}$ for an allyl $-\text{C}-\lambda-\text{C}=\text{C}-\lambda-\text{C}-$ group (Jones, A. P. & Ysard, N. 2025). For comparison the National Institute of Standards and Technology (NIST) Computational Chemistry Comparison and Benchmark DataBase (CCCBDB) bond energies of aromatic C=C in benzene are $5.4\ \text{eV}$, olefinic C=C in cyclohexene are $6.3\ \text{eV}$. The common alkynic C≡C bond energy is around $8.7\ \text{eV}$. Based on these values, the $-\text{C}-\lambda-\text{C}=\text{C}-$ bonds are the most fragile and will be the first to break. Most of these allyl-adjacent groups can be found in the hydrogenated aromatic rings present in the domains. The UV photon-driven fragmentation will generate aromatic units linked by aliphatics. If these domains have a curvature they can form fullerenes by photoinduced dehydrogenation (Irlé et al. 2006; Micelotta et al. 2012). Alternatively, the UV field will break the aliphatic links and release the aromatic domains. If the domains have significant curvature, due to the presence of five-membered rings, they can close and form fullerenes (Berné & Tielens 2012; Berné et al. 2015).

UV processing can further convert the defects (five/-seven rings) into six-membered rings and form PAHs.

Another fragmentation mechanism could involve multi-cation induced Coulomb fragmentation (MCF) due to the repulsion of charges localized on the aromatic domains. Jones, A. P. & Ysard, N. (2025) estimated that molecular particles with radii of 0.2 and 0.25 nm would contain on average two aromatic domains and that each domain would have to carry a charge of $+2/+3$ to induce fragmentation. Our molecular particle has a radius of 0.55 nm and contains six aromatic domains. This could lead to a total charge of $+12$ or more that is higher than the charges found on interstellar grains (Ibáñez-Mejía et al. 2019) and this process is therefore unlikely.

5. CONCLUSIONS

We have selected a potential hydrogenated carbonaceous molecular particle containing 130 carbon atoms, described its structure, computed its emission spectra for the 3.2–3.6 μm , 5–13.7 μm , and 15.0–22.0 μm regions and compared it to the NIRSpec and MIRI/MRS spectra of IRAS 05110. The structure contains aromatic/hydrogenated aromatic domains with some defects (pairs of five/seven aromatic rings) and these domains are linked by aliphatic links. It has some curvature, due to the presence of aromatic five-membered rings, which reduces the particle’s radius. The computed spectra are in qualitative agreement with the observed spectra. We have attributed the broad plateau, previously unassigned, to combinations of aliphatic, aromatic, and olefinic C–H bending modes that are coupled with the collective skeletal C–C modes of the disordered and distorted nature of these molecular particles. The UV photon-driven fragmentation of hydrogenated carbonaceous molecular particles can lead to the formation of PAHs and fullerenes and help explain the life-cycle of carbon in the Universe.

6. ACKNOWLEDGMENTS

This work is in part based on observations made with the NASA/ESA/CSA James Webb Space Telescope. All of the data presented in this article were obtained from the Mikulski Archive for Space Telescopes (MAST) at the Space Telescope Science Institute. These observations are associated with program #4678 (DOI: 10.17909/zs5n-2t21). E.P., J.C., N.C., and C.B. acknowledge support from the Canadian Space Agency (CSA, 24JWGO3B01), and the Natural Sciences and Engineering Research Council of Canada. Support for US investigators (G.C.S., R.S.) in program #4678 was provided by NASA through a grant from

the Space Telescope Science Institute, which is operated by the Association of Universities for Research in Astronomy, Inc., under NASA contract NAS 5-03127. A.R. acknowledges support from the Internal Scientist Funding Model (ISFM) Laboratory Astrophysics Directed Work Package Round 3 at NASA Ames and the High-End Computing (HEC) resources provided by the NASA Advanced Supercomputing (NAS) Division at the NASA Ames Research Center. G.C.S. was funded as part of this observing program by STScI through grant JWST-GO-04678.001-A. R.S.’s contribution to the research described here was carried out at the Jet Propulsion Laboratory, California Institute of Technology, under a contract with NASA (80NM0018D0004). D.A.G.H. acknowledges support from the State Research Agency (AEI) of the Spanish Ministry of Science, Innovation, and Universities (MICIU) of the Government of Spain, and the European Regional Development fund (ERDF), under grant PID2023-147325NB-I00/AEI/10.13039/501100011033. This publication is based upon work from COST Action CA21126 - Carbon molecular nanostructures in space (NanoSpace), supported by COST (European Cooperation in Science and Technology).

REFERENCES

- 527 Acke, B., Bouwman, J., Juhász, A., et al. 2010, *ApJ*, 718,
 528 558, doi: [10.1088/0004-637X/718/1/558](https://doi.org/10.1088/0004-637X/718/1/558)
 529 Bauschlicher, C. W., & Lawson, J. W. 2010, *Chemical*
 530 *Physics*, 374, 77
 531 Bauschlicher, C. W., & Ricca, A. 2010, *Molecular Physics*,
 532 108, 2647, doi: [10.1080/00268976.2010.518979](https://doi.org/10.1080/00268976.2010.518979)
 533 Bauschlicher, Jr., C. W., Ricca, A., Boersma, C., &
 534 Allamandola, L. J. 2018, *ApJS*, 234, 32,
 535 doi: [10.3847/1538-4365/aaa019](https://doi.org/10.3847/1538-4365/aaa019)
 536 Berné, O., Montillaud, J., & Joblin, C. 2015, *A&A*, 577,
 537 A133, doi: [10.1051/0004-6361/201425338](https://doi.org/10.1051/0004-6361/201425338)
 538 Berné, O., & Tielens, A. G. G. M. 2012, *Proceedings of the*
 539 *National Academy of Science*, 109, 401,
 540 doi: [10.1073/pnas.1114207108](https://doi.org/10.1073/pnas.1114207108)
 541 Bhattarai, B., Pandey, A., & Drabold, D. 2018, *Carbon*,
 542 131, 168
 543 Bilek, M. M. M., McKenzie, D. R., McCulloch, D. G., &
 544 Goringe, C. M. 2000, *Phys. Rev. B*, 62, 3071,
 545 doi: [10.1103/PhysRevB.62.3071](https://doi.org/10.1103/PhysRevB.62.3071)
 546 Boersma, C., Bauschlicher, C. W., Allamandola, L. J., et al.
 547 2010, *A&A*, 511, A32, doi: [10.1051/0004-6361/200912714](https://doi.org/10.1051/0004-6361/200912714)
 548 Boersma, C., Bauschlicher, C. W., Ricca, A., et al. 2011,
 549 *The Astrophysical Journal*, 729, 64,
 550 doi: [10.1088/0004-637x/729/1/64](https://doi.org/10.1088/0004-637x/729/1/64)
 551 Boersma, C., Bauschlicher, C. W., Ricca, A., et al. 2014,
 552 *ApJS*, 211, 8, doi: [10.1088/0067-0049/211/1/8](https://doi.org/10.1088/0067-0049/211/1/8)
 553 Clark, N., et al. 2026
 554 Dartois, E., Charon, E., Engrand, C., Pino, T., & Sandt, C.
 555 2020, *A&A*, 637, A82, doi: [10.1051/0004-6361/202037725](https://doi.org/10.1051/0004-6361/202037725)
 556 Demyk, K., Joblin, C., de Bentzmann, L., Toubanc, D., &
 557 Mulas, G. 2026, arXiv preprint arXiv:2607.16018
 558 Frisch, M. J., Pople, J. A., & Binkley, J. S. 1984, *The*
 559 *Journal of Chemical Physics*, 80, 3265
 560 Frisch, M. J., Trucks, G. W., Schlegel, H. B., et al. 2016,
 561 *Gaussian 16 Revision A.03*, Gaussian Inc., Wallingford,
 562 CT
 563 Geballe, T. R., Joblin, C., D'Hendecourt, L. B., et al. 1994,
 564 *ApJL*, 434, L15, doi: [10.1086/187561](https://doi.org/10.1086/187561)
 565 Geballe, T. R., Tielens, A. G. G. M., Kwok, S., & Hrivnak,
 566 B. J. 1992, *ApJL*, 387, L89, doi: [10.1086/186312](https://doi.org/10.1086/186312)
 567 Geballe, T. R., & van der Veen, W. E. C. J. 1990, *A&A*,
 568 235, L9
 569 Ibáñez-Mejía, J. C., Walch, S., Ivlev, A. V., et al. 2019,
 570 *MNRAS*, 485, 1220, doi: [10.1093/mnras/stz207](https://doi.org/10.1093/mnras/stz207)
 571 Irle, S., Zheng, G., Wang, Z., & Morokuma, K. 2006, *The*
 572 *Journal of Physical Chemistry B*, 110, 14531
 573 Jensen, P. A., Shannon, M. J., Peeters, E., Sloan, G. C., &
 574 Stock, D. J. 2022, *A&A*, 665, A153,
 575 doi: [10.1051/0004-6361/202141511](https://doi.org/10.1051/0004-6361/202141511)
 576 Jones, A. P. 2012, *A&A*, 542, A98,
 577 doi: [10.1051/0004-6361/201118483](https://doi.org/10.1051/0004-6361/201118483)
 578 —. 2013, *A&A*, 555, A39,
 579 doi: [10.1051/0004-6361/201321687](https://doi.org/10.1051/0004-6361/201321687)
 580 Jones, A. P., & Ysard, N. 2025, *A&A*, 704, A132,
 581 doi: [10.1051/0004-6361/202555203](https://doi.org/10.1051/0004-6361/202555203)
 582 Kresse, G., & Furthmüller, J. 1996, *Phys. Rev. B*, 54, 11169
 583 Kresse, G., & Furthmüller, J. 1996, *Computational*
 584 *Materials Science*, 6, 15
 585 Kresse, G., & Hafner, J. 1993, *Phys. Rev. B*, 47, 558(R)
 586 —. 1994, *Phys. Rev. B*, 49, 14251
 587 Kresse, G., & Joubert, D. 1999, *Phys. Rev. B*, 59, 1758
 588 Kwok, S., & Zhang, Y. 2011, *Nature*, 479, 80
 589 Kwok, S., & Zhang, Y. 2013, *ApJ*, 771, 5,
 590 doi: [10.1088/0004-637X/771/1/5](https://doi.org/10.1088/0004-637X/771/1/5)
 591 Langhoff, S. R. 1996, *The Journal of Physical Chemistry*,
 592 100, 2819, doi: [10.1021/jp952074g](https://doi.org/10.1021/jp952074g)
 593 Lazzarini, A., Piovano, A., Pellegrini, R., et al. 2016,
 594 *Catalysis Science I& Technology*, 6, 4910,
 595 doi: [10.1039/c6cy00159a](https://doi.org/10.1039/c6cy00159a)
 596 Li, A., & Draine, B. T. 2012, *ApJL*, 760, L35,
 597 doi: [10.1088/2041-8205/760/2/L35](https://doi.org/10.1088/2041-8205/760/2/L35)
 598 Materese, C. K., Bregman, J. D., & Sandford, S. A. 2017,
 599 *The Astrophysical Journal*, 850, 165,
 600 doi: [10.3847/1538-4357/aa960d](https://doi.org/10.3847/1538-4357/aa960d)
 601 Matsuura, M., Bernard-Salas, J., Lloyd Evans, T., et al.
 602 2014, *MNRAS*, 439, 1472, doi: [10.1093/mnras/stt2495](https://doi.org/10.1093/mnras/stt2495)
 603 Micelotta, E. R., Jones, A. P., Cami, J., et al. 2012, *ApJ*,
 604 761, 35, doi: [10.1088/0004-637X/761/1/35](https://doi.org/10.1088/0004-637X/761/1/35)
 605 Papoular, R. 2013, *MNRAS*, 434, 862,
 606 doi: [10.1093/mnras/stt1078](https://doi.org/10.1093/mnras/stt1078)
 607 Peeters, E., Hony, S., Van Kerckhoven, C., et al. 2002,
 608 *A&A*, 390, 1089, doi: [10.1051/0004-6361:20020773](https://doi.org/10.1051/0004-6361:20020773)
 609 Perdew, J. P., & Wang, Y. 1992, *Phys. Rev. B*, 45, 13244
 610 Pino, T., Dartois, E., Cao, A.-T., et al. 2008, *A&A*, 490,
 611 665, doi: [10.1051/0004-6361:200809927](https://doi.org/10.1051/0004-6361:200809927)
 612 Sadjadi, S., Kwok, S., & Zhang, Y. 2016, *Journal of*
 613 *Physics: Conference Series*, 728, 062003,
 614 doi: [10.1088/1742-6596/728/6/062003](https://doi.org/10.1088/1742-6596/728/6/062003)
 615 Sadjadi, S., Zhang, Y., & Kwok, S. 2015, *The Astrophysical*
 616 *Journal*, 801, 34, doi: [10.1088/0004-637X/801/1/34](https://doi.org/10.1088/0004-637X/801/1/34)
 617 —. 2017, *The Astrophysical Journal*, 845,
 618 <https://api.semanticscholar.org/CorpusID:250839423>
 619 Sandford, S. A., Bernstein, M. P., & Materese, C. K. 2013,
 620 *ApJS*, 205, 8, doi: [10.1088/0067-0049/205/1/8](https://doi.org/10.1088/0067-0049/205/1/8)
 621 Sloan, G. C. 2017, *Planet. Space Sci.*, 149, 32,
 622 doi: [10.1016/j.pss.2017.07.017](https://doi.org/10.1016/j.pss.2017.07.017)
 623 Sloan, G. C., Jura, M., Duley, W. W., et al. 2007, *ApJ*,
 624 664, 1144, doi: [10.1086/519236](https://doi.org/10.1086/519236)

- 625 Sloan, G. C., Lagadec, E., Zijlstra, A. A., et al. 2014, *ApJ*,
626 791, 28, doi: [10.1088/0004-637X/791/1/28](https://doi.org/10.1088/0004-637X/791/1/28)
- 627 Sloan, G. C., et al. 2026
- 628 Volk, K., Sloan, G. C., & Kraemer, K. E. 2020,
629 *Astrophysics and Space Science*, 365, 88,
630 doi: [10.1007/s10509-020-03798-2](https://doi.org/10.1007/s10509-020-03798-2)
- 631 Wang, Q.-D. 2017, *Computational and Theoretical*
632 *Chemistry*, 1115, 45,
633 doi: <https://doi.org/10.1016/j.comptc.2017.05.034>
- 634 Yang, X. J., Glaser, R., Li, A., & Zhong, J. X. 2016,
635 *MNRAS*, 462, 1551, doi: [10.1093/mnras/stw1740](https://doi.org/10.1093/mnras/stw1740)